

# Red phosphors for solid-state (LED) lighting applications

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- Funding source: Mitsubishi Chemical Center for Advanced Materials



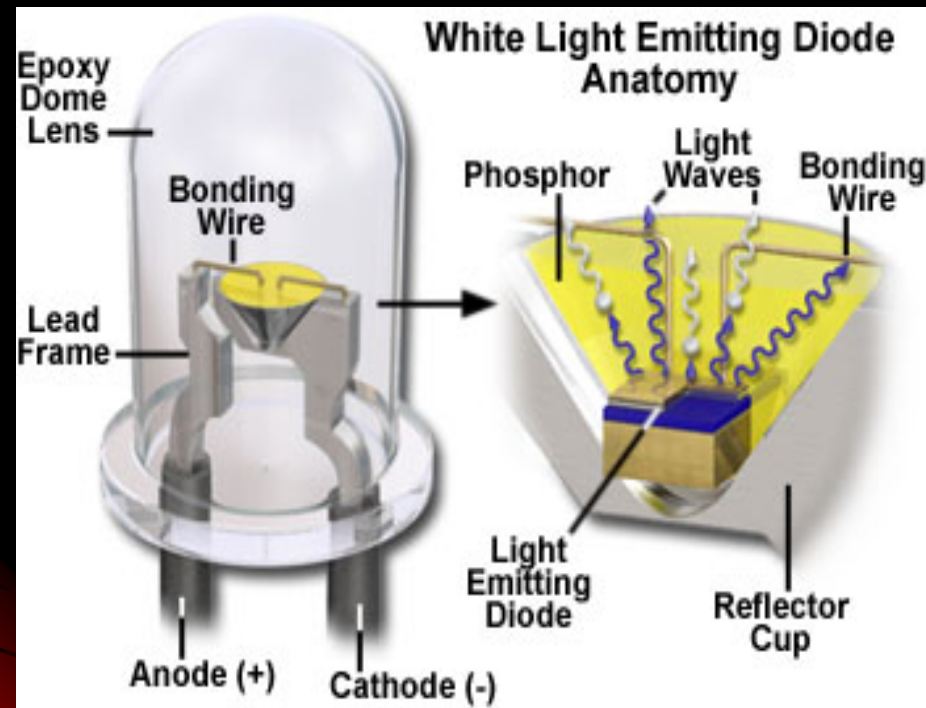
# Introduction

- Solid state lighting (LEDs) use about an eighth of the energy that incandescent bulbs use, have a lifespan of over 50,000 hours and do not contain the mercury found in fluorescent bulbs.
- This translates to a 5+ year lifespan and energy savings of over 85% for lighting.
- Applications of LEDs:



# Research objectives

- A phosphor is a solid/powder that can absorb radiation of one wavelength and convert it to light of another color.

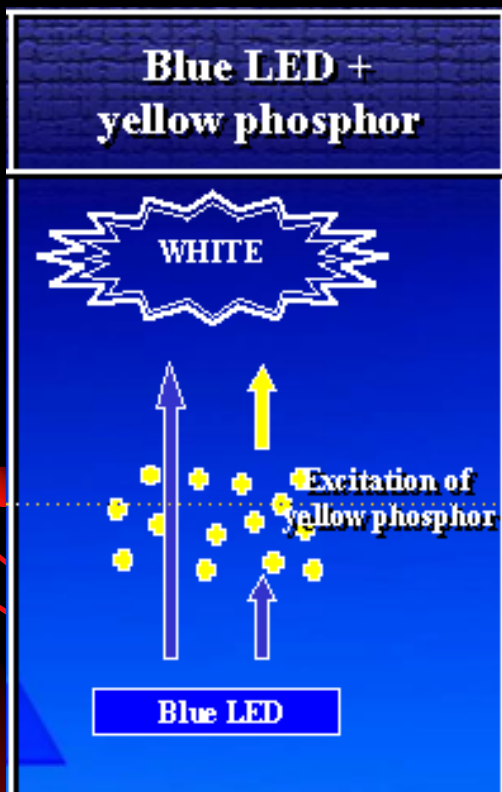


[www.physics.utoledo.edu](http://www.physics.utoledo.edu)



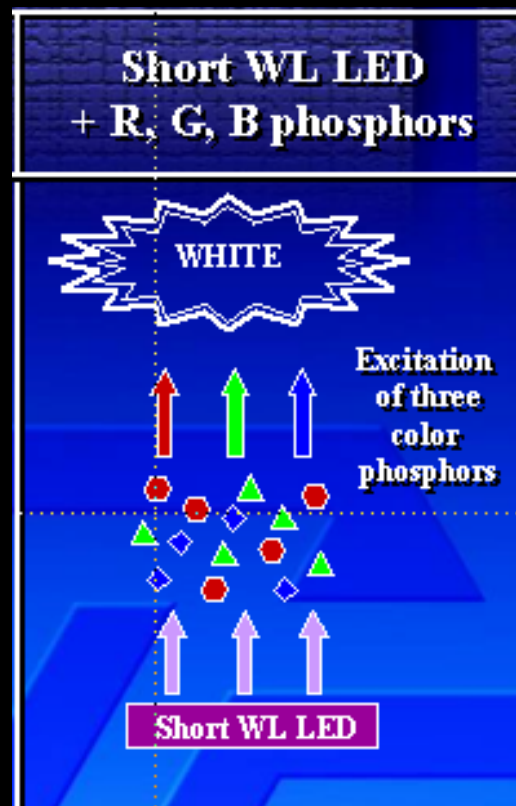
[www.phosphortech.com](http://www.phosphortech.com)

- Our current LEDs have a blue chip (460 nm) that excites a yellow phosphor, emitting a white light.
- The next generation of LEDs will use an ultraviolet chip (400 nm) that will excite red, green and blue phosphors.



- High luminous intensity
- Easy manufacturing (less handling materials)

[www.toyoda-gosei.com](http://www.toyoda-gosei.com)



- RGB+P Regions = Good CRI
- High Color Saturation
- Excellent Color Stability

# Research objectives...continued

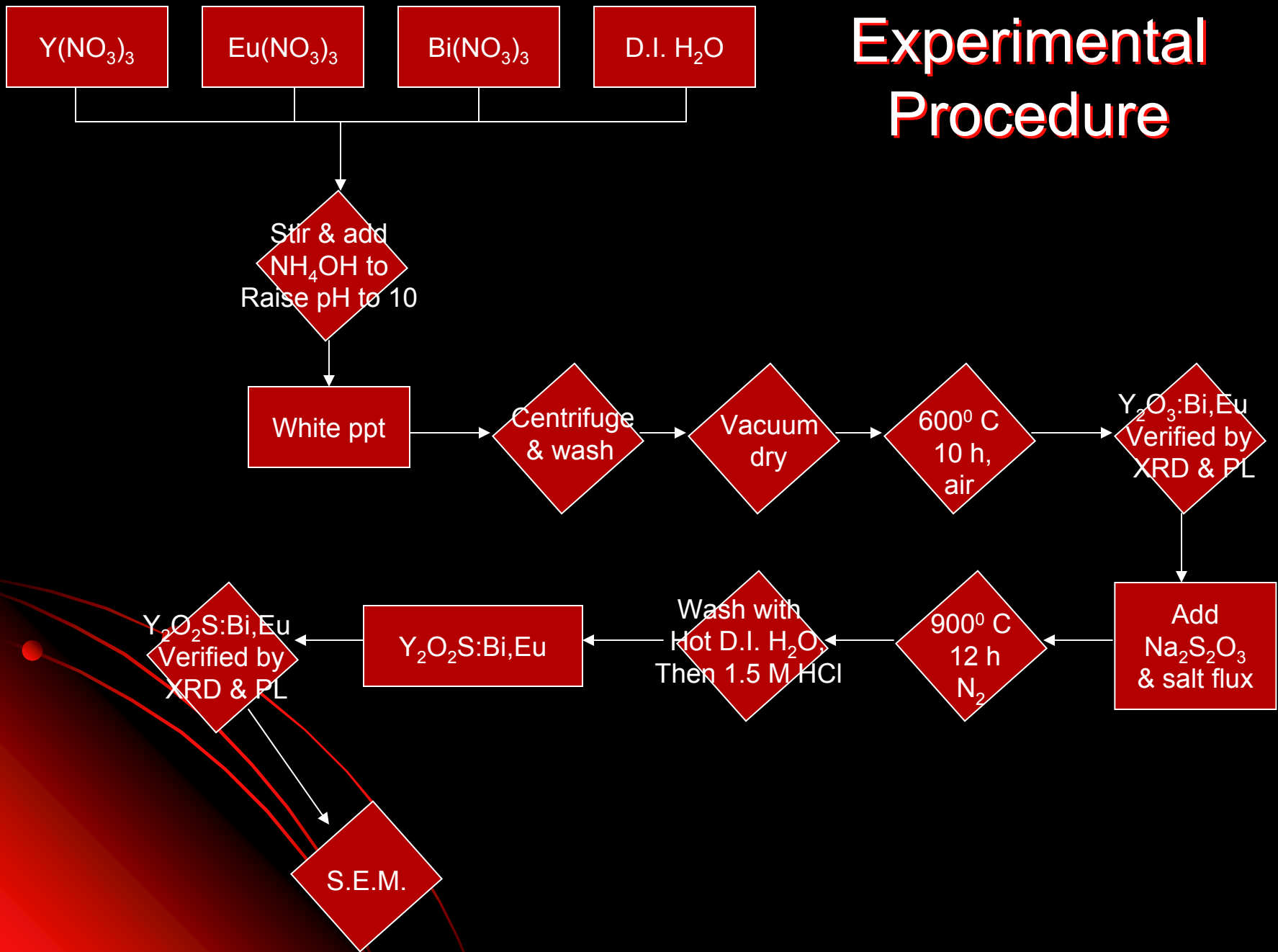
Commercial red phosphor- Yttrium oxysulfide  $\text{Y}_2\text{O}_2\text{S}:\text{Eu}$

Excitation wavelength 340 nm, emission wavelength 627 nm

Can we tune the  
particle size by  
use of salt flux?

Can addition of  $\text{Bi}^{3+}$  shift  
excitation to a longer  
wavelength?

# Experimental Procedure





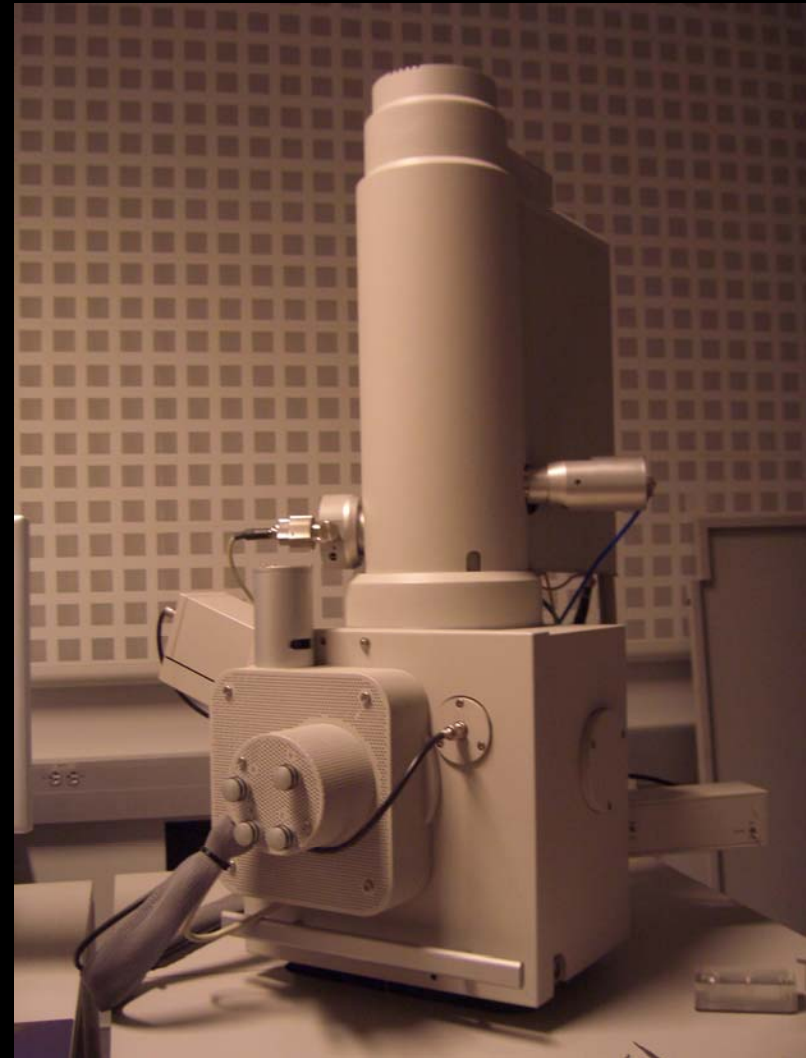
# Equipment used



Tube furnace (up to 1200°C)



Gold sputtering  
(for SEM)

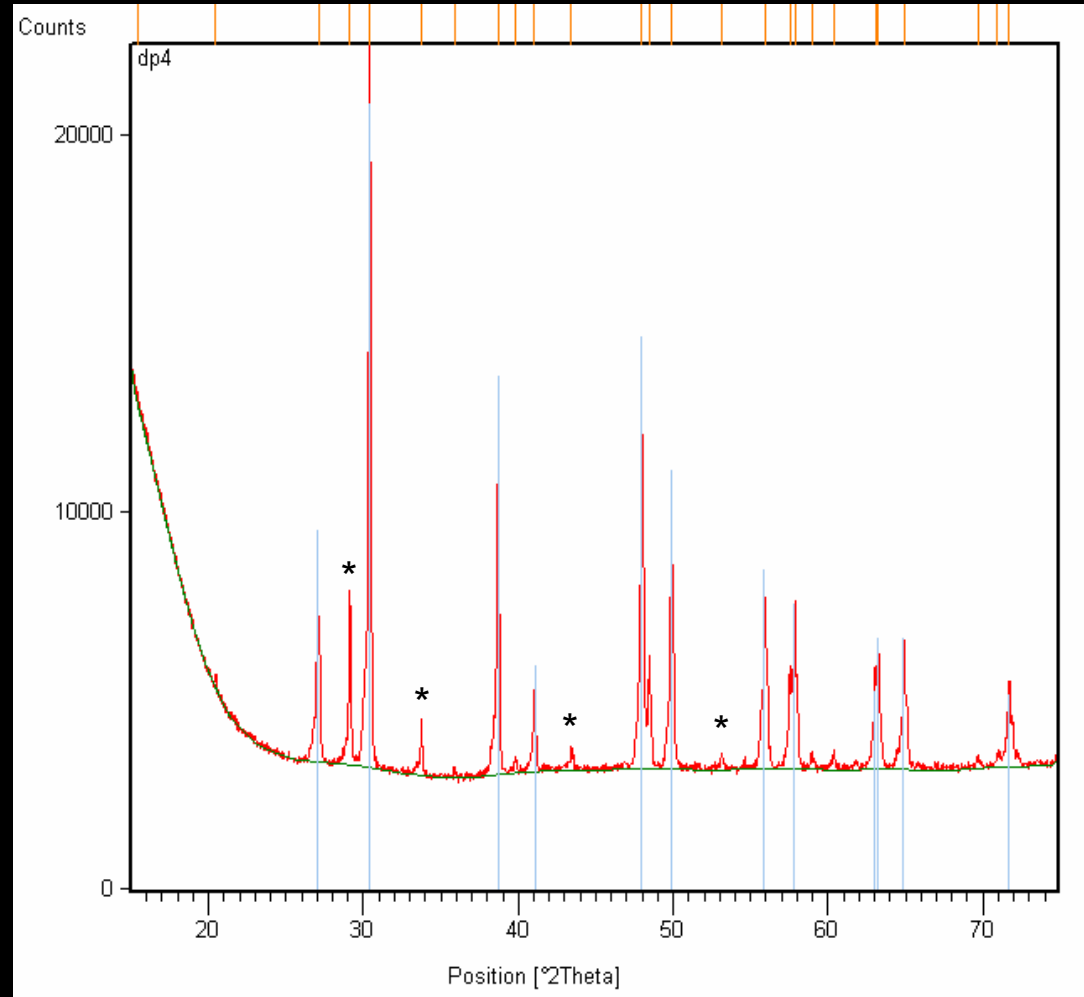


Scanning electron microscope (SEM)

# Characterization – X-ray diffraction

To find out if the sample synthesized was pure

- Mainly pure  $Y_2O_2S$   
(compared to standard)
- Impurity of  $Y_2O_3$   
(~30%)

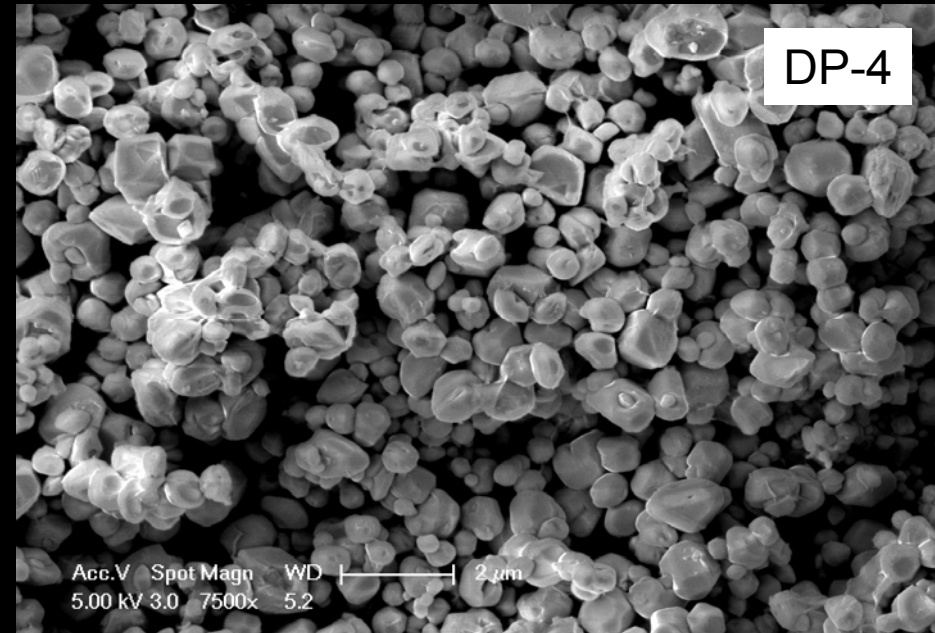
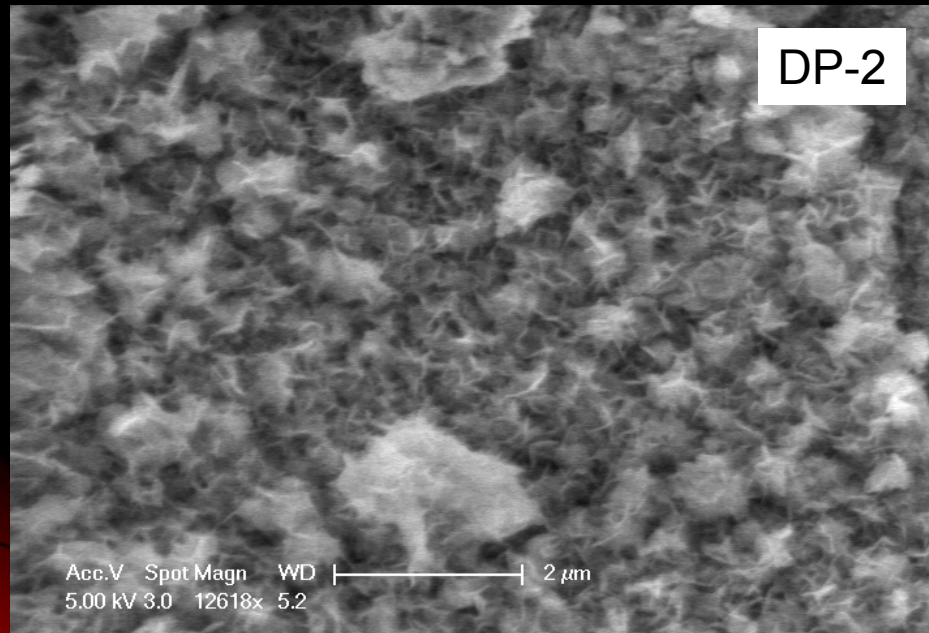


XRD pattern of  $Y_2O_2S:Eu(5\%)$  without salt flux.



# Characterization – Scanning Electron Microscopy (SEM)

To find out the morphology



Reactant  $Y_2O_3$

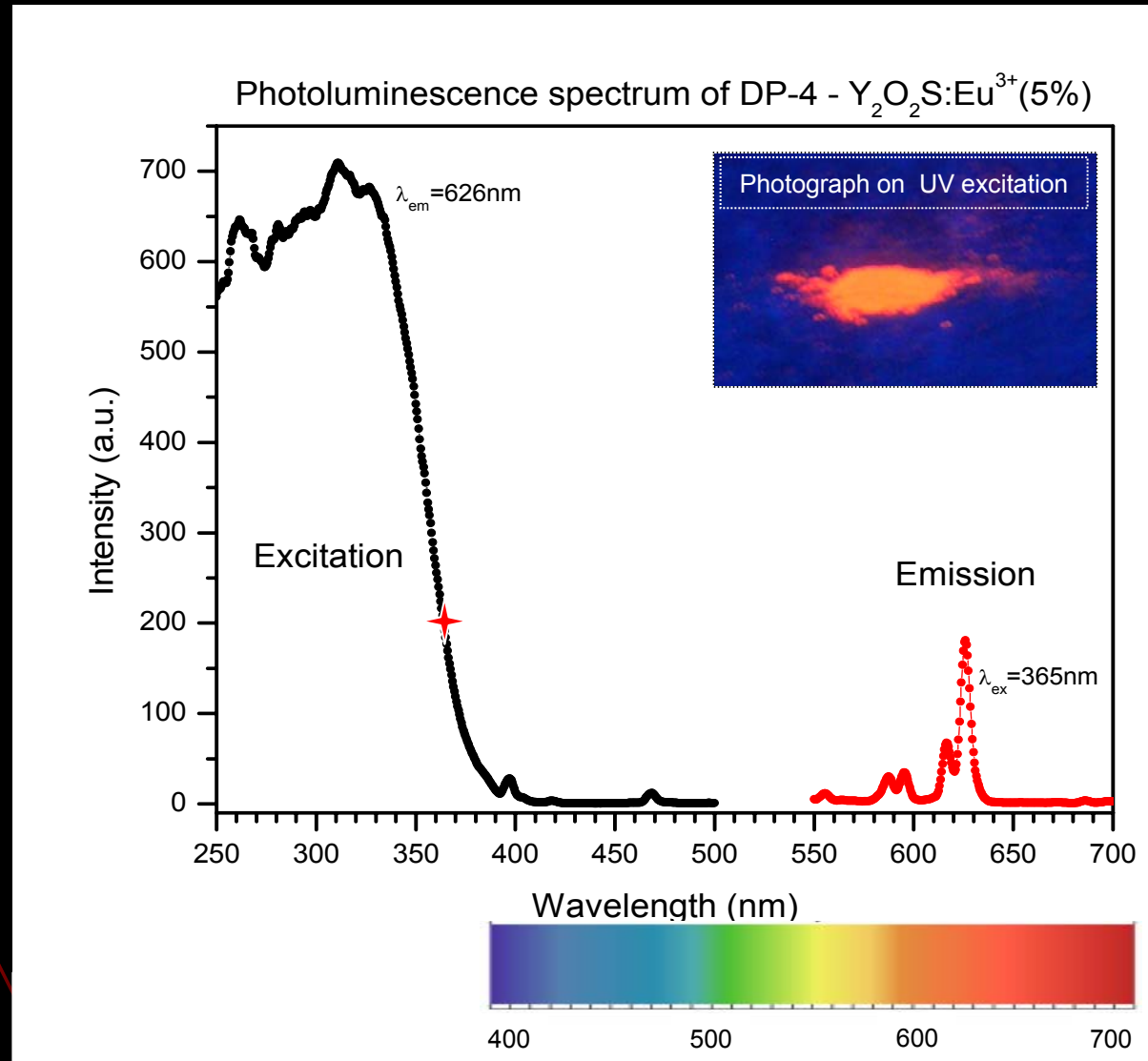
Product  $Y_2O_2S$  (no salt flux)

- Products look totally different from reactants.
- Product contains particles having a size between 200-1000 nm.

# Characterization – Photoluminescence Spectroscopy

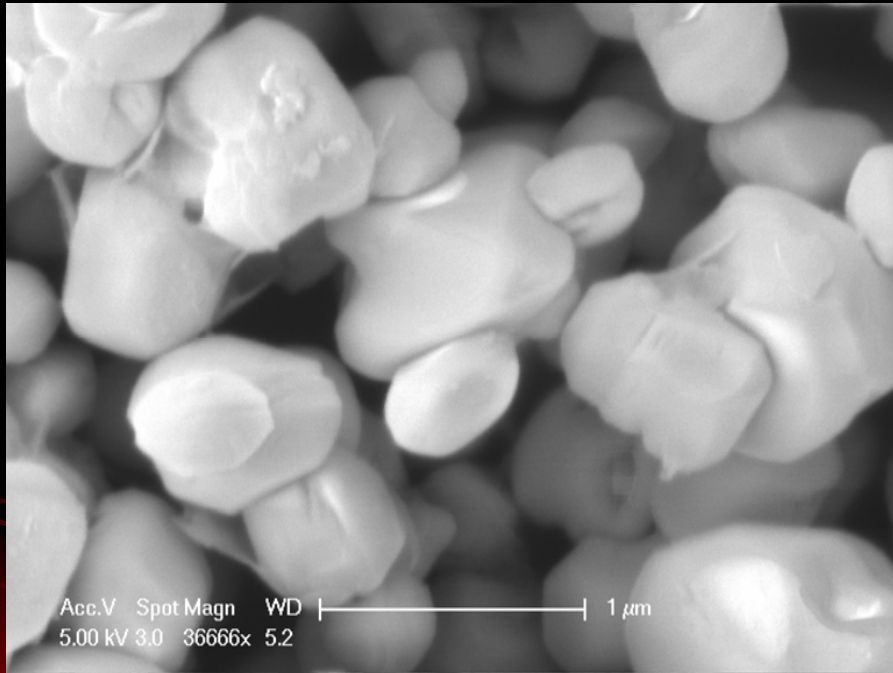
To measure luminescence (determine excitation and emission)

- Broad excitation with maximum at 330 nm.
- Line emission with maximum at 626 nm.



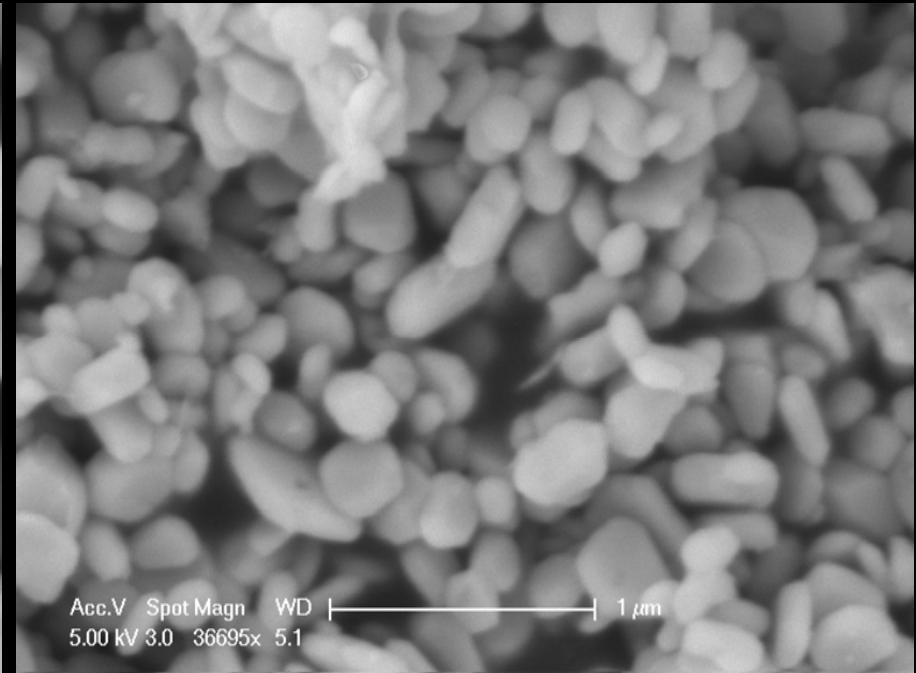
# Effect of salt-flux addition

No salt flux



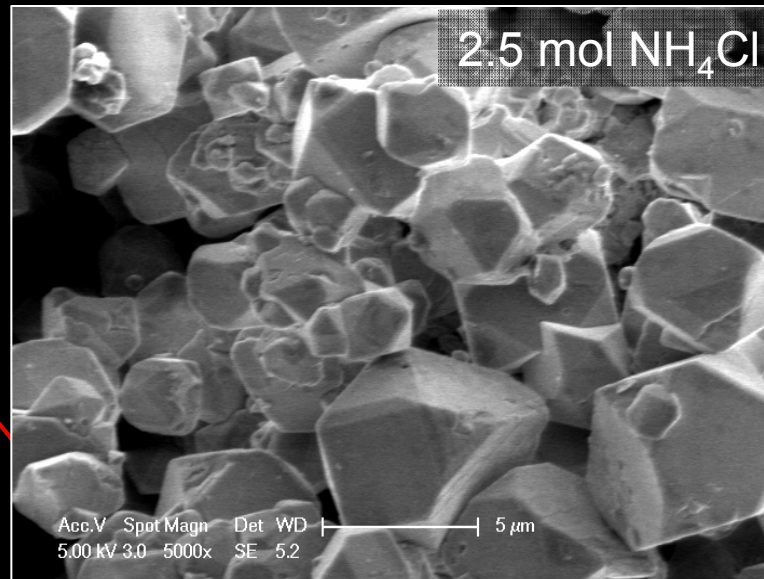
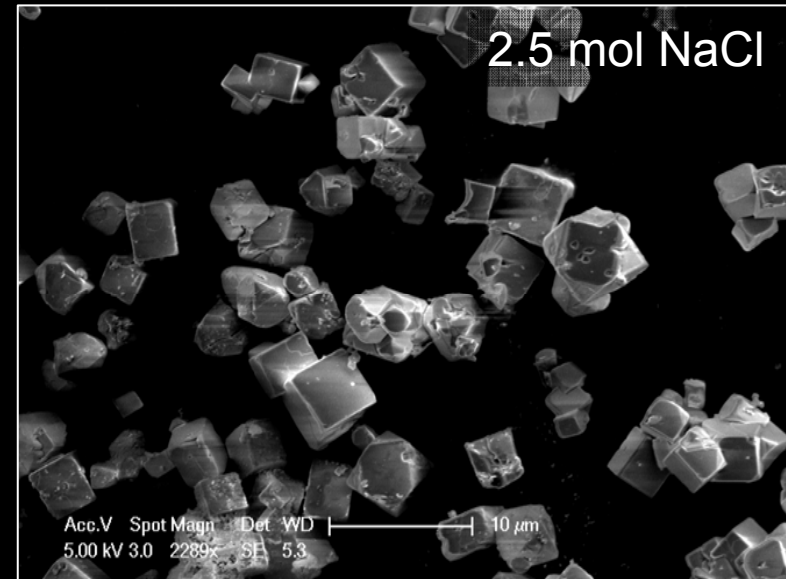
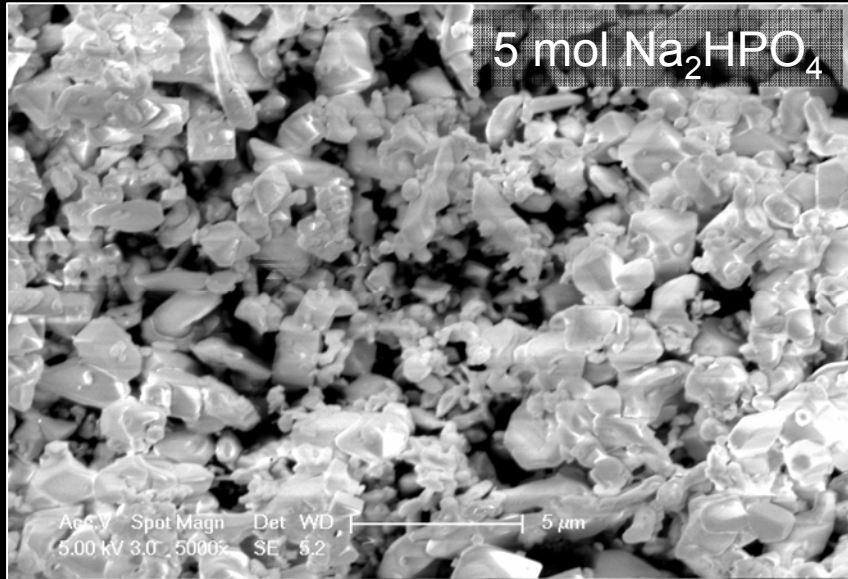
- Particle size 200-1000 nm

2.8 moles of  $\text{Na}_2\text{HPO}_4$



- Particle size 100-400 nm
- Slightly elongated rods present
- Further characterization under progress

# Effects of adding different salt-fluxes



Further characterization under progress



## Summary

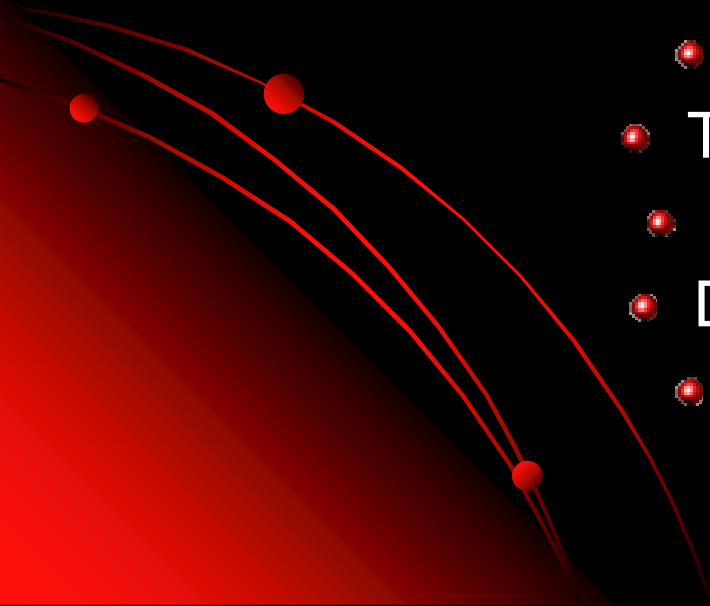
- Optimized the synthesis of  $Y_2O_2S$  phosphor.
- Addition of salt-fluxes such as  $Na_2HPO_4$ ,  $KCl$ ,  $NaCl$ ,  $NH_4Cl$  was studied.
- $Bi^{3+}$  doping did not shift the excitation of  $Y_2O_2S:Eu$ .
- Learned the use of different instruments such as centrifuges, furnaces, ovens etc. to synthesize the samples and XRD, PL, SEM to characterize them.

## Future Plans

- Optimize the flux and reaction conditions to obtain uniform morphology.
- Perform optical studies on the samples.



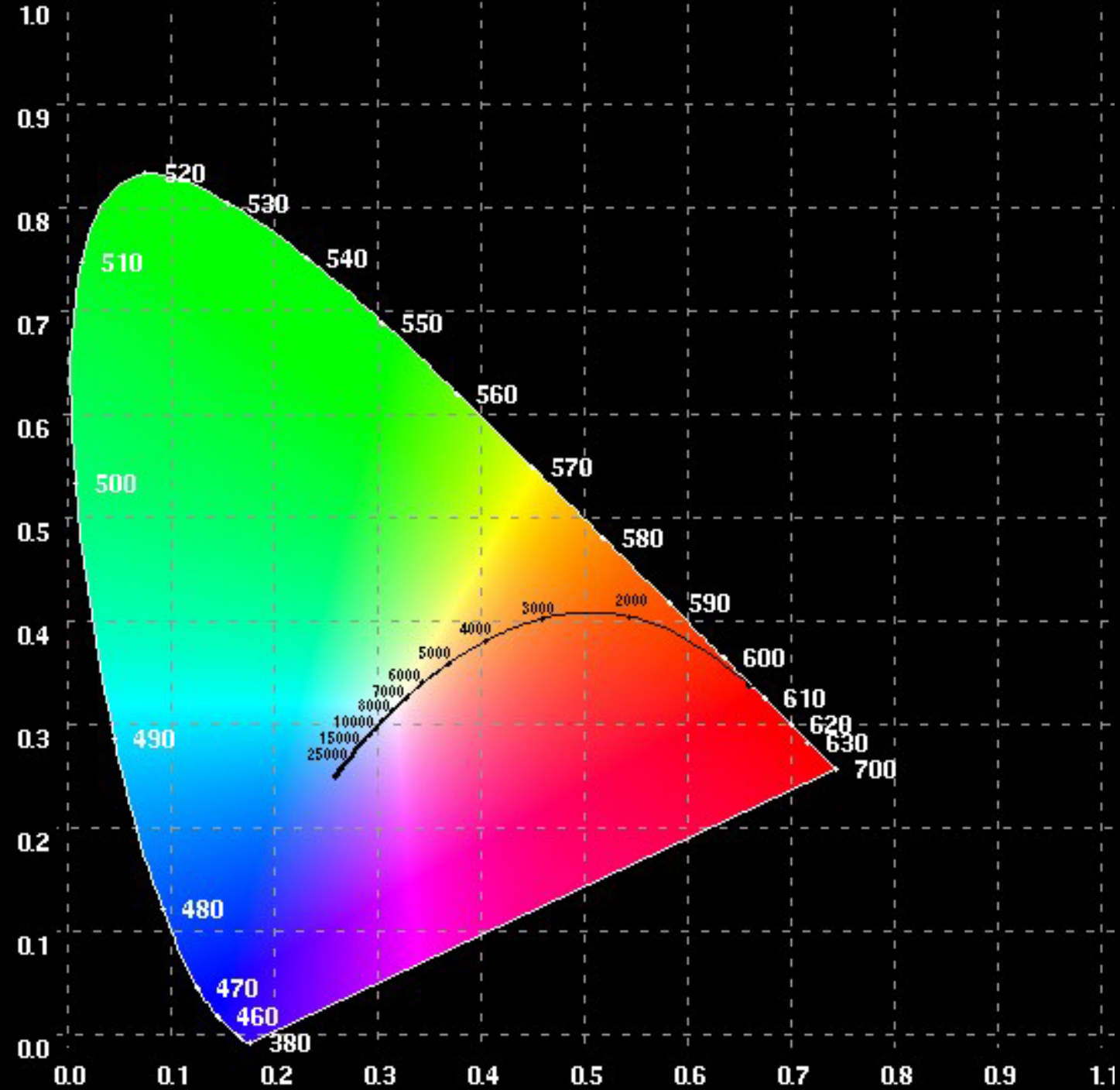
# Acknowledgements

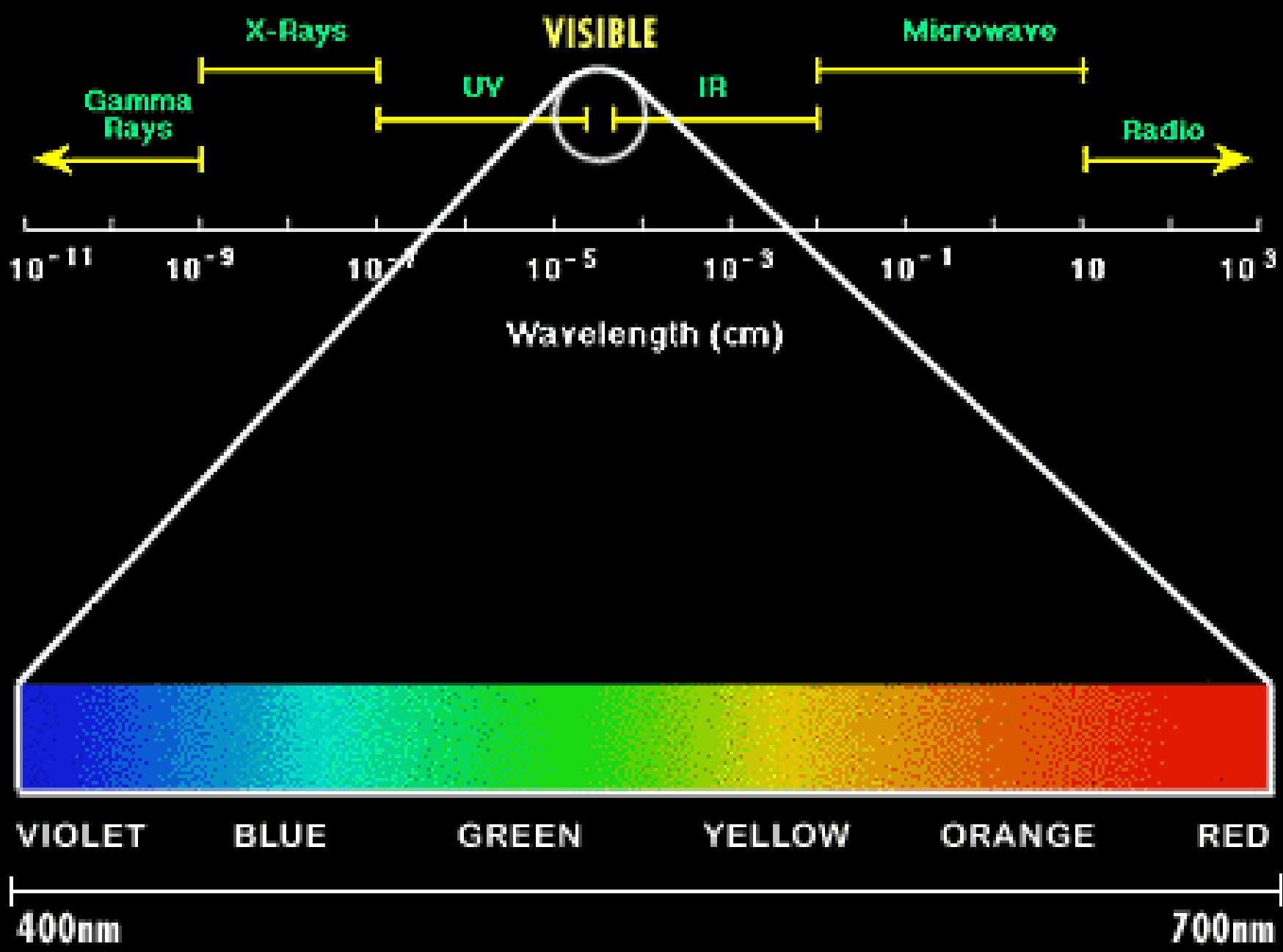
- Dr. Nick Arnold
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  - Liu-Yen Kramer
  - Prof. Evelyn Hu
  - Lad Handelman, Linda Seals  
Lady & Blossom
  - Hanna Pagenkopf
  - The Cheetham Group
  - Dr. Leah Appelhans
  - Dr. Francois Chevire
  - INSET colleagues
- 

A scenic photograph of a sunset over a large body of water, likely a bay or harbor. The sky is filled with vibrant orange and red clouds, with the sun low on the horizon. The water reflects the colors of the sky. In the background, there are dark silhouettes of mountains. The foreground is dark, showing the silhouettes of trees and buildings, with some small lights visible.

Thank You,

Questions?





# THE ELECTROMAGNETIC SPECTRUM

