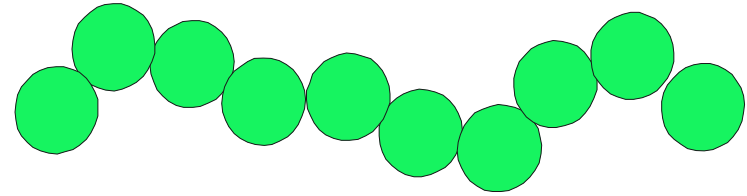


# Diblock Copolymer Reinforced Interface



**Pedro I**

**Mentor: Jaso**

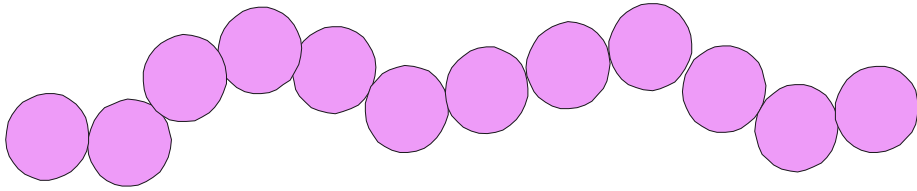
**Professor: E**

**Funding Source: Nation**

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Allan Hancoo

# es between Polystyrene and Polyethylene



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**ional Science Foundation**

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# **Diblock Copolymer Reinforced Interfaces between Polystyrene and Polyethylene**

**Pedro Flores,**

**Allan Hancock College Student**

**Mentor: Jason Benkoski,**

**Professor: E.D. Kramer**

**UCSB Materials Department**

**Funding Source: National Science Foundation**

# Abstract

The interface between Polystyrene (PS) and polyethylene (PE) normally has low fracture energy ( $G_c$ ) on the order of  $1 \text{ J/m}^2$ . The strength of the interface can be improved by reinforcing them with PS-PE diblock copolymer. The areal chain density of the diblock copolymer was held constant for all tests. We measured  $G_c$  as a function of temperature using the asymmetric double cantilever beam test (ADCB). By observing the dependence of  $G_c$  on temperature, we can discuss whether or not the pullout of the PE block from bulk PE is a thermally activated process. Understanding PE fracture properties will help us improve the durability of polymers blends. Combining the two results in the properties shown below.

## **Polyethylene (PE):**

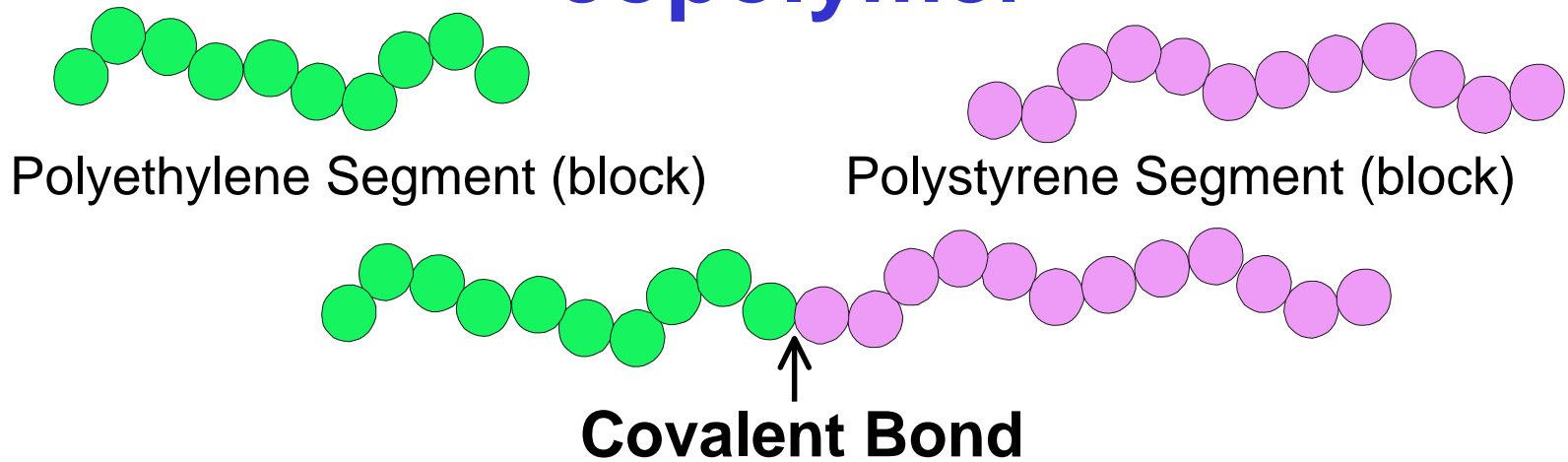
- Flexible, **Tough**
- Used for: Liners, Bullet Proof Vests, Food Packaging

## **Polystyrene (PS):**

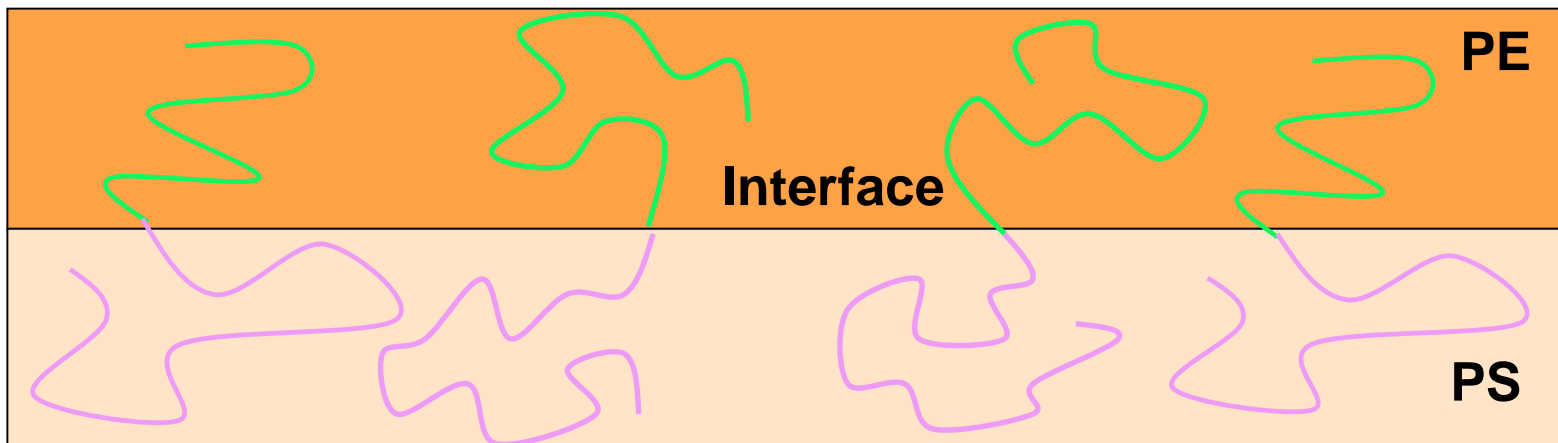
- Hard**, brittle
- Used for: CD Cases, Disposables

**PS + PE = Hard + Tough**

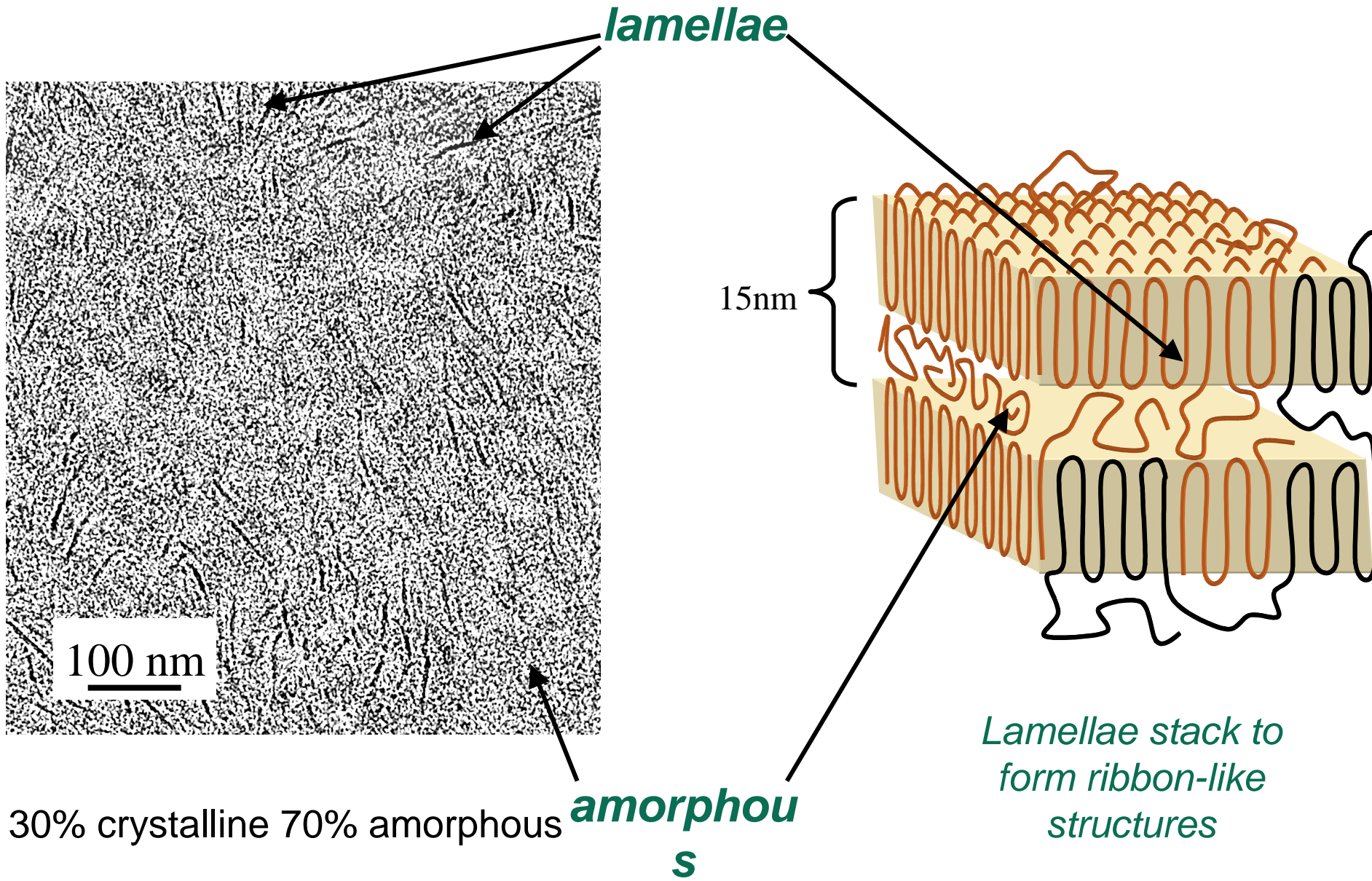
# The polyethylene and polystyrene diblock copolymer



Diblock copolymer is placed in between polyethylene (PE) and polystyrene homopolymers (PS).



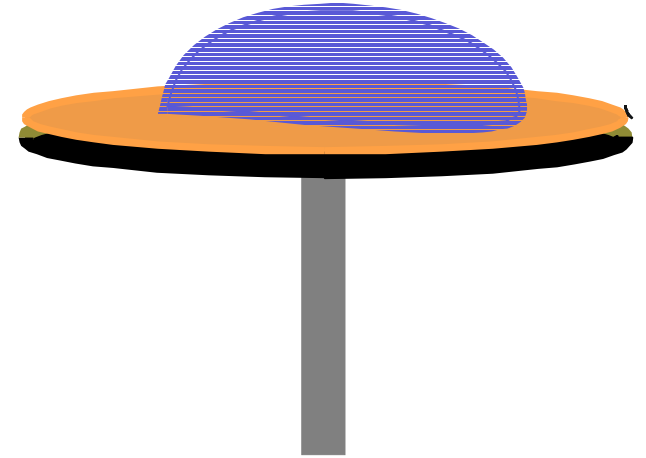
# Semicrystalline Polymers



# Spin Coating Diblock Copolymer on top of the PE Film

## In the Solution:

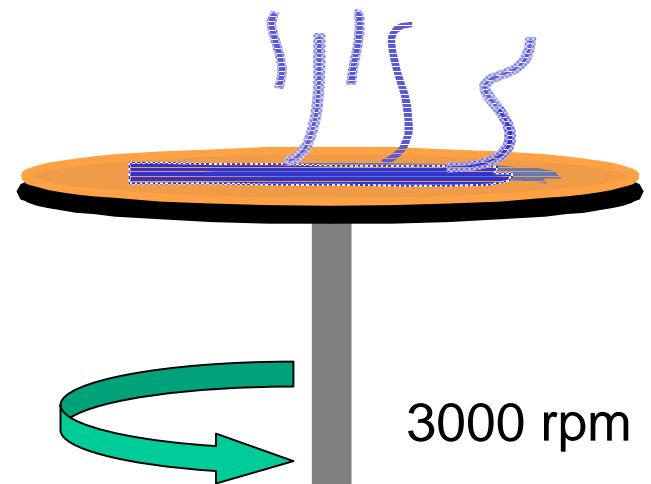
- Toluene(solvent)
- Poly(styrene-b-ethylene)



The solvent evaporates leaving a thin coating on the PE.

## Spin coating both sides of the film:

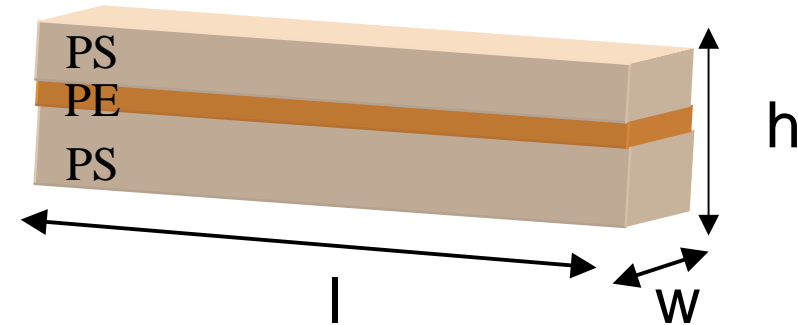
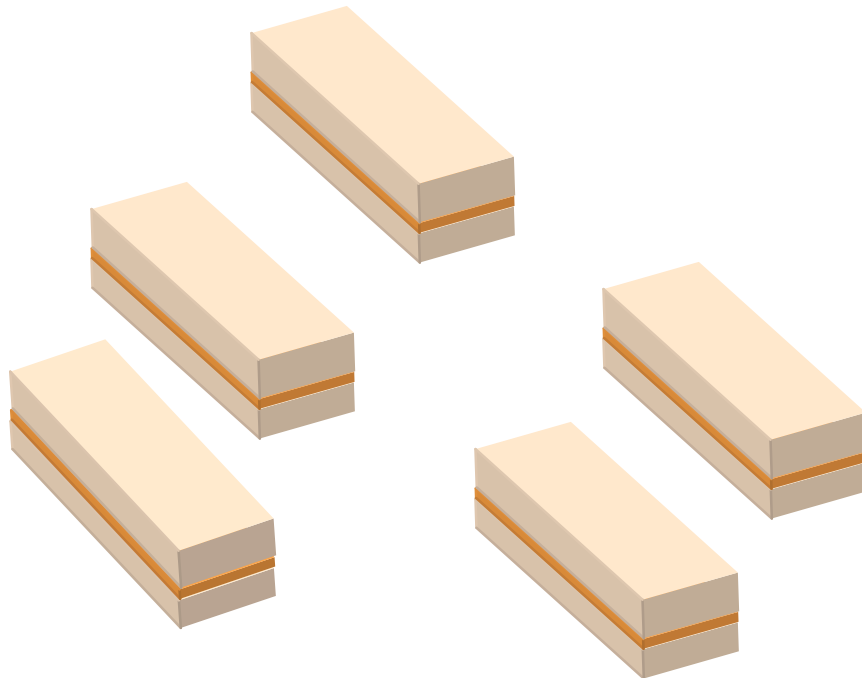
- PS-PE (40,000g/mol-30,000g/mol)
- PS-PE (40,000g/mol-100,000g/mol)
- Areal chain density of 0.2 chains/nm<sup>2</sup>



# Dimensions of Samples

1. Polymers are annealed at 160°C for two hours.
2. Cooled to room temperature in 3 minutes and cut into samples

- PE film is 70  $\mu\text{m}$  thick
- Top PS beam is 2 mm
- Bottom PS beam is 2.5 mm



- Width ( $w$ ) = 8 mm
- Length ( $l$ ) = 40 mm
- Height ( $h$ ) = 4.5 mm

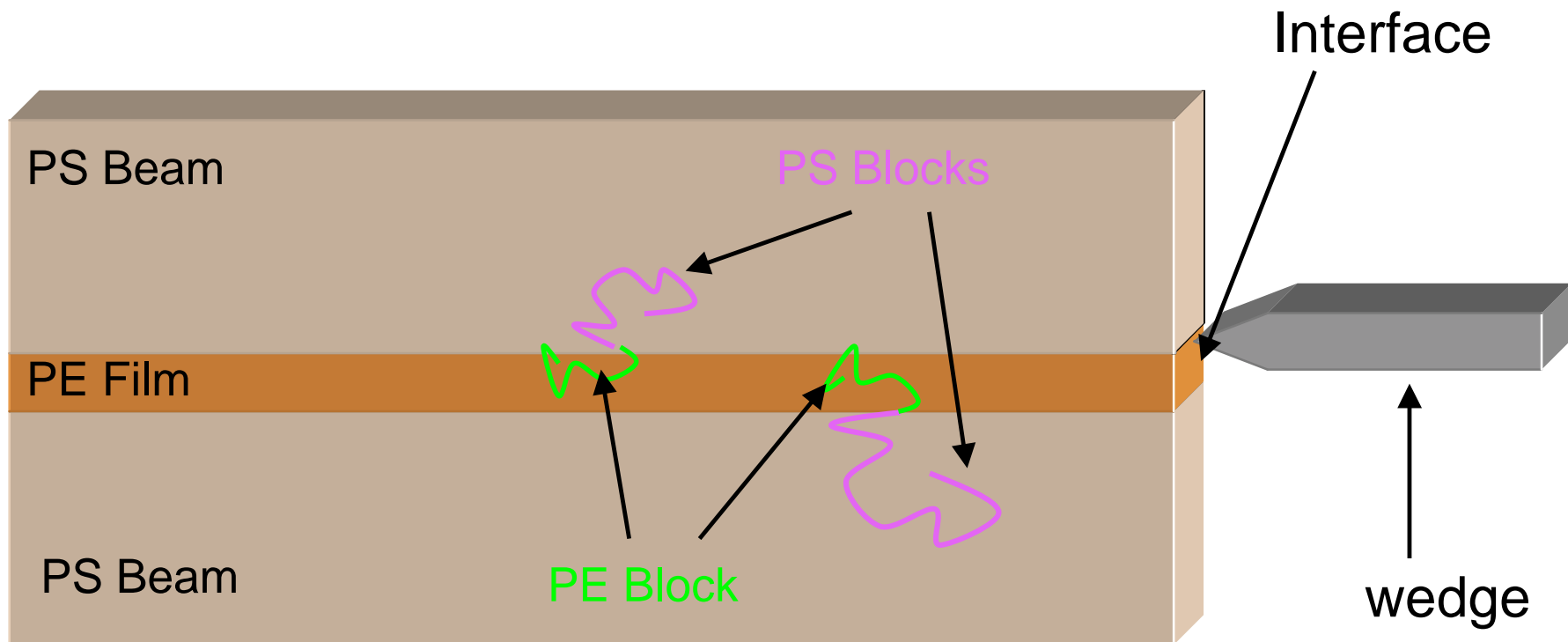


# Asymmetric Double Cantilever Technique

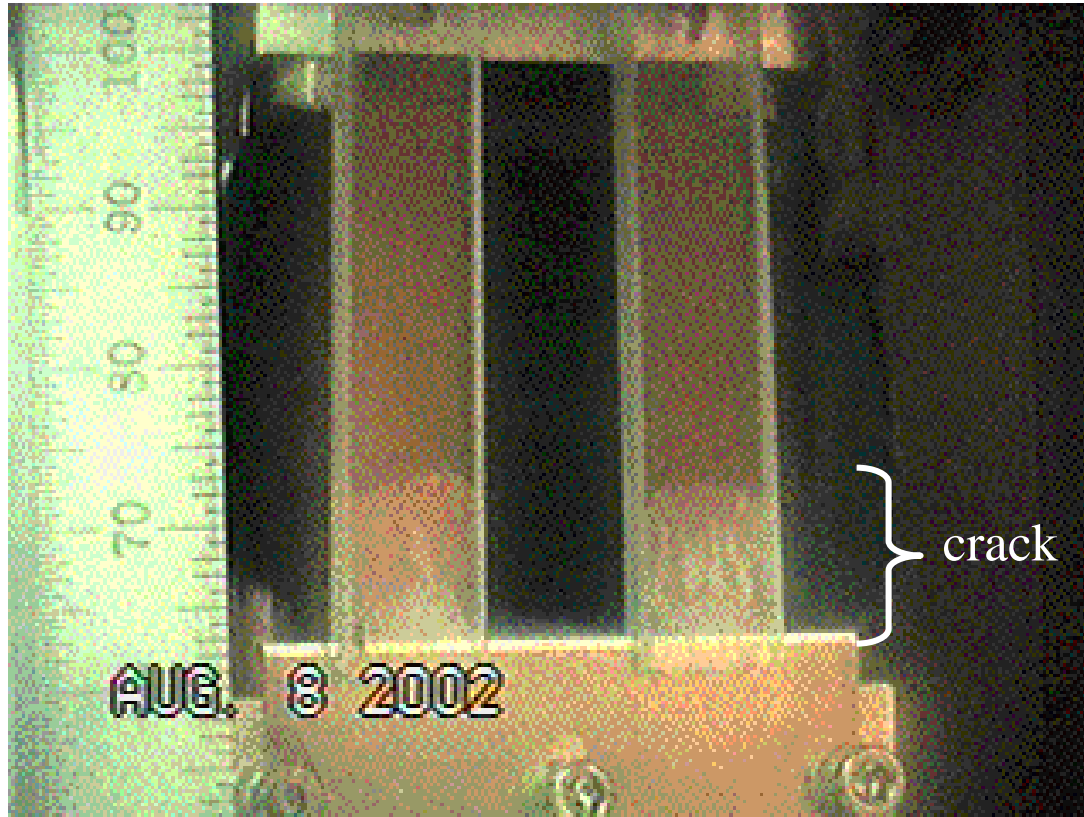
1. 2.7 mm wedge moves through the interface at constant speed.

2. Images are captured every 3 minutes to a computer.

3. Crack lengths are measure using the NIH Image program.

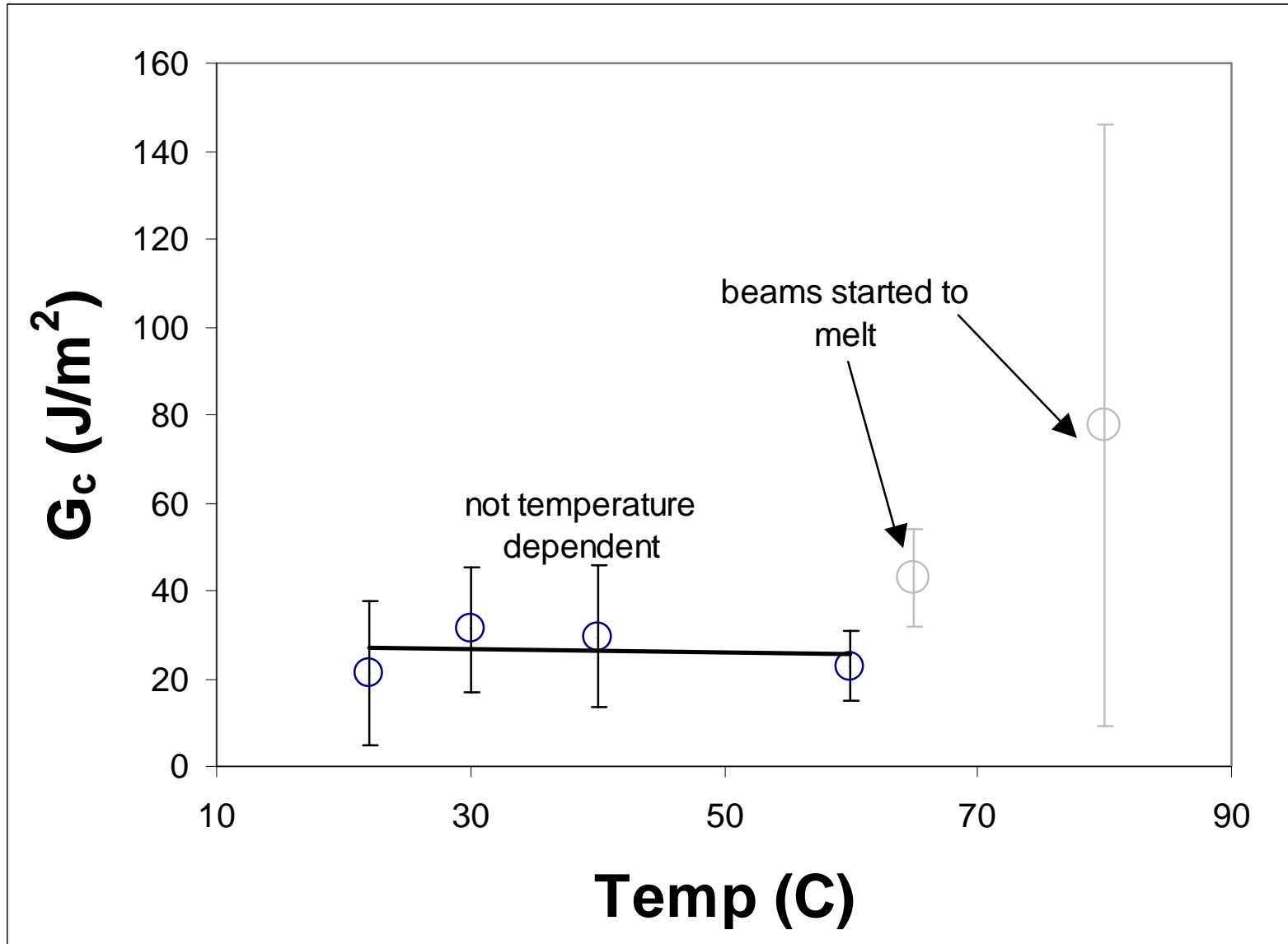


# Fracture Energy Measurements

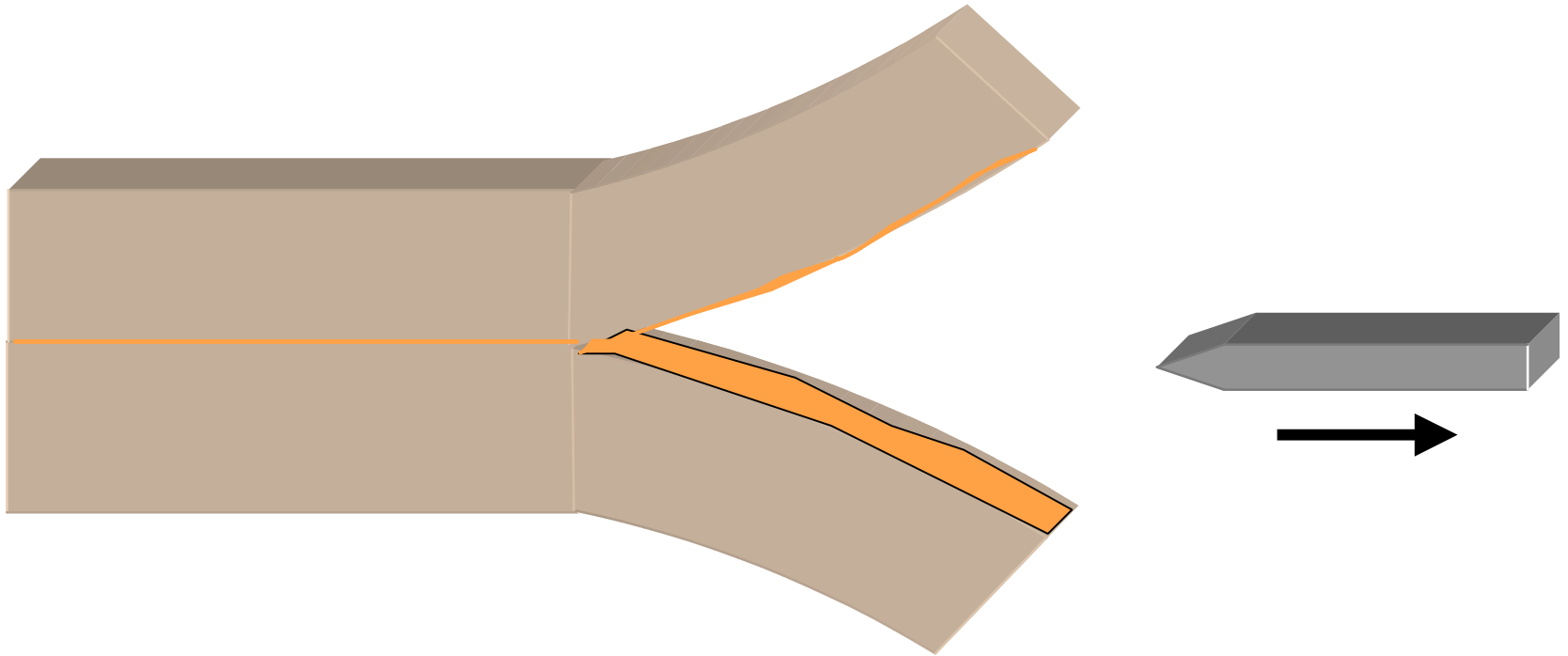


- 3 measurements per picture
  - 30 pictures per sample
  - 6 samples
- **540** measurements per data point

# Fracture Energy vs. Temperature



# Plastic Deformation Above 60°C



- Beams remained bent after blade was removed.
- Too close to  $T_g$  for polystyrene beams

# Fracture Failure

## Pullout



**Occurs when one block is pulled out from its parent homopolymer during fracture.**

**This is what we see for our system.**

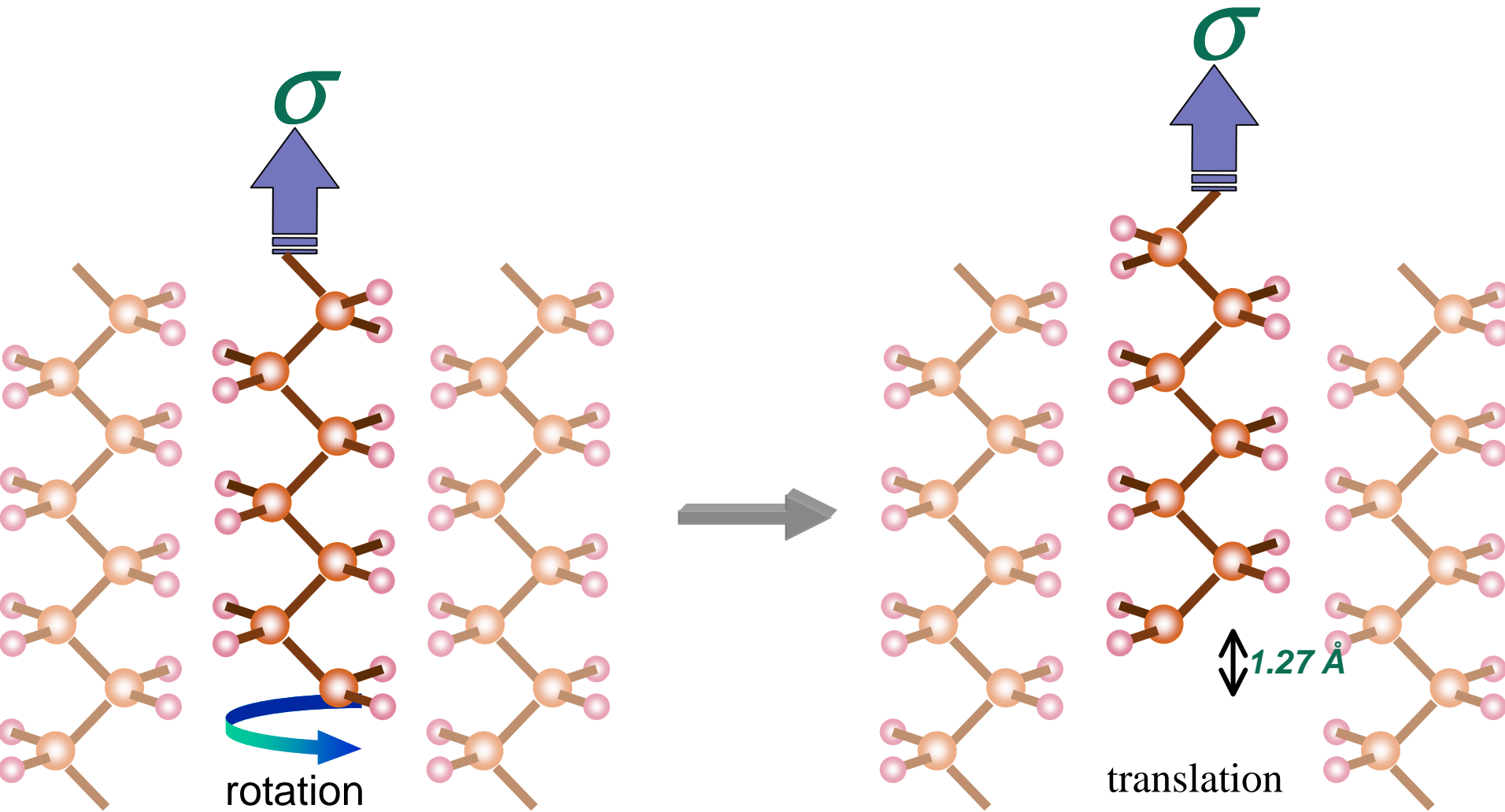
## Scission



Occurs when the diblock copolymer breaks somewhere along its length.

Only at higher molecular weights

# Thermally Activated (Diffusion) Process



**Does pullout involve stress-assisted diffusion?**

# Conclusions

- The pullout of PE blocks do not appear to be thermally activated
  - $G_c$  does not change with temperature
  - Temperature range may be too small
- Cannot make measurements above  $60^\circ\text{C}$ 
  - Beams plastically deform
  - Too close to  $T_g$